

PRENYLFLAVANOLS FROM *TEPHROSIA QUERCETORUM**

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Key Word Index—*Tephrosia quercetorum*; Leguminosae; roots; new prenylflavanols.

Abstract—The roots of *Tephrosia quercetorum* afforded three new flavonoids named quercetols A, B and C. Their structures were established by spectroscopic methods, mainly ^1H NMR.

INTRODUCTION

Previous studies of various Mexican *Tephrosia* species (Leguminosae) have provided a number of novel flavonoids [1–4]. We have now undertaken a study of a new species, *T. quercetorum*, a herb endemic to the southeast of México.

RESULTS AND DISCUSSION

Extraction of the roots of *Tephrosia quercetorum* with petrol, petrol–ethyl acetate (1:1) and methanol, followed in each case by CC and prep. TLC over silica gel (see Experimental), gave three new flavanols quercetols A(1), B(2), and C(3). In addition, the known compounds 5-hydroxy-7-methoxy-8-prenylflavanone (4), [1], 5,7-dimethoxy-8-prenylflavanone (4a) [5], obovatol methyl ether (5) [6], tephrowatsin A (6) [3], and methylhidgartol B (7a) [8], were isolated.

Quercetol A(1) analysed for $\text{C}_{21}\text{H}_{22}\text{O}_5$ (M^+ 354). Its IR spectrum showed the presence of hydroxyl (3440 cm^{-1}) but not carbonyl groups. The UV spectrum suggested a flavan-3,4-diol structure [9]. The structure of 1 was deduced from the ^1H NMR spectrum (Table 1) which was similar to that of tephrobottin (7), previously isolated from *Tephrosia abbottiae* [7]. Quercetol A(1) differs from 7 by the presence of an extra hydroxyl group at C-3. According to the ^1H NMR, the two-proton multiplet at $\delta 2.15$ (H-3) was replaced by a one-proton signal at $\delta 3.92$ ($J=10, 8, 4\text{ Hz}$), which collapsed to a doublet of doublets upon D_2O addition. Moreover the H-4 signal appeared as a broad doublet at $\delta 4.95$ ($J=4\text{ Hz}$) which was also sharpened on D_2O addition and there was an H-2 signal as a doublet at $\delta 4.92$ ($J=10\text{ Hz}$). A broad singlet at $\delta 2.80$ and a doublet at $\delta 2.65$ ($J=8\text{ Hz}$) were assigned to the hydroxyl protons at C-3 and C-4, respectively, since these signals were interchangeable with D_2O . As in tephrobottin (7), the MS showed a base peak at m/z 219 due to the $[\text{A}_1 - \text{Me}]^+$ fragment and a molecular ion at m/z 354, which confirmed the presence

of an extra oxygen in the molecule. Acetylation of 1 afforded the corresponding diacetate 1a. The ^1H NMR of 1a clearly exhibited two sharp acetate peaks at $\delta 1.80$ and 2.15 and a downfield shift of the H-3 and H-4 signals. It is important to point out that all the flavan-3,4-diols of natural origin, whose structure and stereochemistry have been fully characterized, have been isolated from the wood or bark of *Acacia* species (Leguminosae) [10].

Quercetol B (2) was isolated as a colourless oil. The UV and IR spectra indicated the presence of an unconjugated aromatic system with no carbonyl and hydroxyl group. The structure of 2 followed from the MS and ^1H NMR. The MS displayed a molecular ion and a base peak at m/z 368 $[\text{M}]^+$ and 249 $[\text{A}_1 - \text{Me}]^+$, which indicated that 2 differs from 6 by 14 mass units. The ^1H NMR spectrum of 2 closely resembled that of tephrowatsin A(6) previously isolated from *T. watsoniana* [3], except for the appearance of a peak for an extra methoxyl group at $\delta 3.46$ and the different chemical shift associated with the replacement of the C-4 hydroxyl group by a methoxyl group. Accordingly, the H-4 signal located at $\delta 5.0$ in tephrowatsin A(6) was shifted upfield to 4.54 ppm.

The IR and the UV spectra of quercetol C suggested a hydroxyflavanone structure. The ^1H NMR confirmed the above assumption, since it showed the characteristic ABX signals of the flavanone nucleus at $\delta 5.40$ (H-2) and 2.95 (H-3). The structure of 3 clearly followed from the ^1H NMR which was very similar to that of 5,7-dimethoxy-8-prenylflavanone (4a). Quercetol C(3) differs from 4a by the presence of tertiary hydroxyl groups and two vinyl protons in the side chain. Accordingly the ^1H NMR of 3 showed the following differences: the two vinyl methyl broad singlets ($\delta 1.62$) of 4a were replaced by a pair of overlapping singlets at $\delta 1.35$. A pair of downfield doublets ($\delta 6.55$ and 6.75 , $J=16\text{ Hz}$) characteristic of a *trans* double bond appeared instead of the typical signals for the prenyl group. The MS showed a molecular ion at m/z 368 together with peaks at m/z 353 $[\text{M} - \text{Me}]^+$, 350 $[\text{M} - \text{H}_2\text{O}]^+$, 249 $[\text{A}_1 - \text{H}_2\text{O}]^+$, 91 and 77 which are consistent with the structure 3. The negative value of the optical rotation of compound 3 indicated the absolute configuration 'S' at C-2, [11].

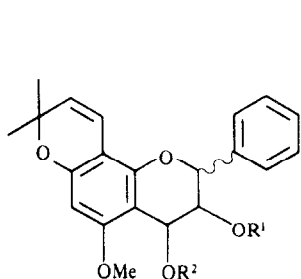
Finally, Jones oxidation of tephrowatsin (A) (6) afforded 8 as an orange solid, mp $114\text{--}116^\circ$; it analysed for $\text{C}_{13}\text{H}_{18}\text{O}_3$ (M^+ 222). Its IR spectrum demonstrated the

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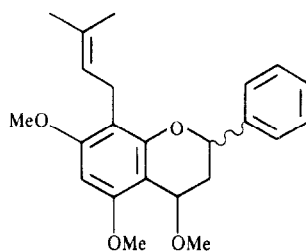
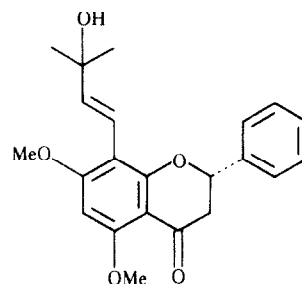
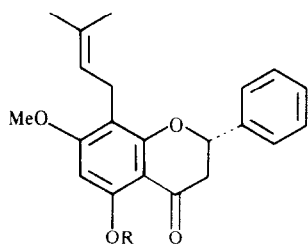
Table 1. ^1H NMR data* of flavanols 1, 2 and 3

H	1	2	3	8
2	4.92 <i>d</i> (10)	5.29 <i>dd</i> (4,12)	5.4 <i>dd</i> (5,11)	
3	3.92 <i>ddd</i> (10, 8, 4)	2.34 <i>m</i>	2.95 <i>m</i>	
4	4.95 <i>d</i> (4)	4.54 <i>t</i> (3)		5.8 <i>s</i>
6	6.03 <i>s</i>	6.12 <i>s</i>	6.09	5.8 <i>s</i>
— ϕ	7.41 <i>m</i>	7.39 <i>m</i>	7.35 <i>m</i>	
7'	6.52 <i>d</i> (10)	3.26 <i>d</i> (7)	6.75 <i>d</i> (16)	3.13 <i>d</i> (7)
8'	5.34 <i>d</i> (10)	5.1 <i>t</i> (7)	6.55 <i>d</i> (16)	5.13 <i>t</i> (7)
<i>gem</i> -Me	1.35 <i>s</i>	1.62 <i>s</i>	1.35 <i>s</i>	1.6 <i>s</i>
—OMe	3.8 <i>s</i>	3.46 <i>s</i> 3.81 <i>s</i> 3.85 <i>s</i>	3.9 <i>s</i> 3.9 <i>s</i> 1.35 <i>s</i>	1.72 <i>s</i> 3.76 <i>s</i> 3.76 <i>s</i>
—OH				6.65 <i>s</i>

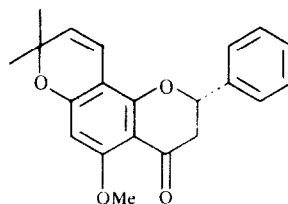
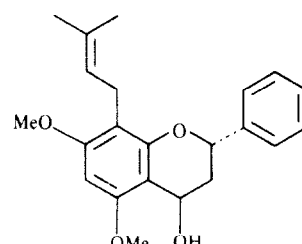
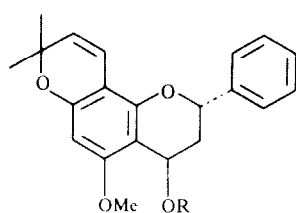
* Run at 80 MHz in CDCl_3 with TMS as internal standard. Values are in ppm (δ). Values in parentheses are coupling constants in Hz.



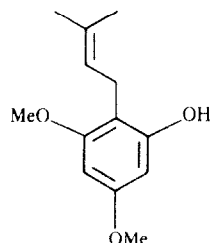
1 $\text{R}^1 = \text{R}^2 = \text{H}$
1a $\text{R}^1 = \text{R}^2 = \text{Ac}$

**2****3**

4 $\text{R} = \text{H}$
4a $\text{R} = \text{Me}$

**5****6**

7 $\text{R} = \text{H}$
7a $\text{R} = \text{Me}$

**8**

presence of an hydroxyl group (3270 cm^{-1}), an olefinic double bond (1675 cm^{-1}) and aromatic double bonds ($1630, 1610\text{ cm}^{-1}$). The $^1\text{H NMR}$ indicated the presence of two aromatic methoxy groups at $\delta 3.76$ (6H), two aromatic protons at $\delta 5.8$ (2H), one phenolic hydroxy group at $\delta 6.65$ and the typical signals for the prenyl group (see Table 1). Based on the above data, **8** must be the 2-prenyl-3,5-dimethoxyphenol derived from the A-ring of tephrowatsin A (**6**) by decarboxylation of the corresponding acid.

EXPERIMENTAL

Mps: uncorr. *Tephrosia quercetorum* Wood was collected in Guerrero, México, ca 21 km from Taxco, during December 1982. A voucher is on deposit at the Herbarium of Instituto de Biología (UNAM), México.

Extraction. The air-dried roots (600 g) of *T. quercetorum* were coarsely powdered and extracted successively with petrol, petrol– Me_2CO (1:1) and MeOH. After evapn of solvents, the green syrups A (27.8 g), B (12.5 g) and C (23.9 g) respectively, were obtained. The petrol extract A (27.8 g) was chromatographed over 450 g silica gel using petrol and mixtures of petrol–EtOAc as eluants. From the fraction eluted with petrol, 5-hydroxy-7-methoxy-8-prenylflavanone (**4**) (2.3 g) was obtained. Fractions eluted with petrol–EtOAc (8:2) afforded 5-methylbovatins (**5**) (100 mg), tephrobottin (**7**) (520 mg) and quercetol A (150 mg). The petrol– Me_2CO (1:1) extract B (12.5 g) was chromatographed over 75 g silica gel using petrol and mixtures of petrol– Me_2CO . From the fractions eluted with petrol 5-hydroxy-7-methoxy-8-prenylflavanone (**4**) (205 mg) and tephrowatsin A (**6**) (35 mg) were obtained. Fractions eluted with petrol– Me_2CO afforded tephrobottin (**7**) (80 mg), 5,7-dimethoxy-8-prenylflavanone (**4a**) (30 mg), quercetol A (**1**) (45 mg) and quercetol C (**3**) (120 mg). In the same way, extract C (23.9 g) afforded 5-hydroxy-7-methoxy-8-prenylflavanone (**4**) (150 mg), tephrobottin (**7**) (35 mg), 5-methylbovatins (**5**) (20 mg), 5,7-dimethyl-8-prenylflavanone (**4a**) (18 mg), methylhydrogattol B (**7a**) (30 mg) and quercetol B (**2**) (35 mg).

Quercetol A (1). $\text{C}_{21}\text{H}_{22}\text{O}_5$ colourless needles, mp $99\text{--}100^\circ$. $[\alpha]_D -2.71^\circ$ (CHCl_3 , c 0.221 g/100 ml). UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 203 (11 078), 234 (32 253), 286 (41 142). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3440, 1585, 1490. EIMS (probe) 70 eV, m/z (rel. int.): 354 $[\text{M}]^+$ (20), 339 $[\text{M}-15]^+$ (47), 321 $[\text{M}-15-18]^+$ (31), 291 (100), 120 (5).

Quercetol A diacetate (1a). Acetylation of **1** (59 mg) with Ac_2O –pyridine gave, after prep. TLC, the diacetate **1a**. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 204 (20 106), 235 (27 660), 287 (6085); IR $\nu_{\text{max}}^{\text{film}}$ cm^{-1} : 1745, 1620, 1585, 1490, EIMS (probe) 70 eV, m/z (rel. int.): 438 $[\text{M}]^+$ (16), 423 $[\text{M}-\text{Me}]^+$ (71), 381 $[\text{M}-\text{C}_2\text{H}_5\text{O}]^+$ (5), 321 $[\text{M}-\text{C}_2\text{H}_4\text{O}]^+$ (20), 219 (100).

Quercetol B (2). $\text{C}_{23}\text{H}_{28}\text{O}_4$, colourless oil, $[\alpha]_D -42.18^\circ$ (CHCl_3 , c 0.192). UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 211 (66240), 277 (1648). IR $\nu_{\text{max}}^{\text{film}}$ cm^{-1} : 1610, 1500. EIMS (probe) 70 eV, m/z (rel. int.): 368 $[\text{M}]^+$ (70), 337 $[\text{M}-\text{OMe}]^+$ (47), 353 $[\text{M}-\text{Me}]^+$ (4), 249 $[\text{M}-\text{C}_8\text{H}_8]^+$ (100), 221 $[\text{M}-\text{CO}]^+$ (75).

Quercetol C (3). $\text{C}_{22}\text{H}_{24}\text{O}_5$, colourless needles, mp $198\text{--}200^\circ$, $[\alpha]_D -66^\circ$ (CHCl_3 , c 0.1). UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 207 (20 392), 263 (27 416), 283 (12 482), 337 (3006). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3430, 1660, 1590, 1560, 790. EIMS (probe) 70 eV, m/z (rel. int.): 368 $[\text{M}]^+$ (26), 350 $[\text{M}-\text{H}_2\text{O}]^+$ (15), 353 $[\text{M}-\text{Me}]^+$ (10), 249 (47), 205 (100), 104 (10), 91 (13), 77 (18).

Oxidation of tephrowatsin A (6). Jones oxidation of **6** (8 mg) gave, after prep. TLC the phenol **8** (8 mg), IR $\nu_{\text{max}}^{\text{film}}$ cm^{-1} : 3270, 1675, 1630, 1610, EIMS (probe) 70 eV, m/z (rel. int.): 222 $[\text{M}]^+$ (8), 207 $[\text{M}-\text{Me}]^+$ (100), 192 $[\text{M}-2\text{Me}]^+$ (29), 69 (C_5H_4) (15).

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